

METABOLIC DISORDERS OF ACID-BASE CHEMISTRY

Disorder	Etiology	Pathophysiology	Lab Findings and Diagnosis	Treatment
METABOLIC ALKALOSIS	<p style="text-align: center;">Generation</p> <p>LOSS OF PLASMA H⁺</p> <p>Renal</p> <p style="padding-left: 20px;"><i>Increased aldosterone</i> Sodium-dependent: electroneutrality Sodium-independent: activate apical H⁺ ATPase in intercalated cells)</p> <p style="padding-left: 20px;"><i>Diuretics (except CA inhibitors)</i> <i>Bartter and Gitelman</i> <i>Primary Hyperaldosteronism</i> <i>Milk-Alkali Syndrome</i> <i>Post-Hypercapneic Alkalosis</i></p> <p>GI</p> <p style="padding-left: 20px;"><i>Emesis and NG suction</i> <i>Factitious Diarrhea (loss of Cl⁻)</i> <i>Adenoma</i></p> <p>Tracellular H⁺ Shift Occurs in hypokalemia</p> <p>GAIN OF PLASMA HCO₃⁻</p> <p>Bicarbonate Infusion</p> <p>Massive transfusion (citrate)</p> <p>CONTRACTION ALKALOSIS</p> <p>Loss of bicarbonate-free fluid (e.g. diuresis of edema fluid)</p> <p style="text-align: center;">Maintenance</p> <p>INTRAVASCULAR VOLUME DEPLETION</p> <p style="padding-left: 20px;">Activation of RAAS: AngII → increase PCT H⁺ excretion via activation of Na⁺/H⁺ exchangers</p> <p style="padding-left: 20px;">Secondary hyperaldosteronism</p>	<p>Urinalysis shows characteristic chemistries</p> <p>EARLY PHASE (1 – 3 d)</p> <p>Increased bicarbonate Increased Na⁺ Increased K⁺ Decreased Cl⁻ pH > 6.5</p> <p>CHRONIC PHASE (> 3 d)</p> <p>Decreased bicarbonate Decreased Na⁺ Decreased K⁺ Decreased Cl⁻ pH < 5.5: paradoxical aciduria</p> <p>The kidney protects ECF volume by massive reabsorption of all presented ions!</p> <p>Thus, metabolic alkalosis generally is NOT self-limiting</p>	<p>The most important test: Urine Cl⁻</p> <p>Urine Cl⁻ < 15 mEq/L Upper GI loss Remote diuretic therapy Post-hypercapneic hypokalemia</p> <p>Tx: volume resuscitation with NML saline + K⁺ replacement (KCl)</p> <p>Urine Cl⁻ > 15 mEq/L Primary hyperaldosteronism Steroids Bartter Syndrome</p> <p>Tx: treat underlying disease</p>	<p>Treat underlying disease + infuse NML saline for volume expansion</p>

	<p>CHLORIDE DEPLETION Increased CD HCO₃⁻ reabsorption Involves Type A intercalated cells: low luminal Cl⁻ → increased apical secretion via Cl⁻/H⁺ cotransporter Decreased CD HCO₃⁻ secretion Involves Type B intercalated cells: low luminal Cl⁻ → decreased activity of Cl⁻/HCO₃⁻ exchanger (less Cl⁻ reabsorption).</p> <p>HYPOKALEMIA Increase in ammoniogenesis (PCT) via intracellular acidosis → increased secretion of organic acids</p>			
<p>METABOLIC ACIDOSIS</p>	<p>NON-ANION GAP (NAGMA) Loss of bicarbonate Diarrhea (also results in hypokalemia) GI fistula Type II RTA (proximal RTA) Decreased regeneration of bicarbonate Type II RTA (distal hyperkalemic) Type IV RTA (distal hypokalemic) Primary hypoaldosteronism Hyperkalemia</p> <p>ANION GAP (AGMA) High Acid Input Poisoning: EtOH, MeOH, paraldehydes, salicylates, ethylene glycol Shock and lactic acidosis Ketoacidosis (DKA, starvation, alcoholism) Low Acid Output Renal failure</p> <p>MIXED METABOLIC ACIDOSIS</p>		<p>Diagnosis of RTAs</p> <p>Type I RTA Distal hypokalemic Decreased H⁺ ATPase activity in the CD Etiology: AR and AD, amphotericin B Serum HCO₃⁻ < 10 mEq/L Urine pH > 5.6 always</p> <p>Type II RTA Proximal hypokalemic Decreased HCO₃⁻ reabsorption in the PCT (e.g. defective serosal Na⁺/HCO₃⁻ cotransporter) Urine can be acidified Serum HCO₃⁻ = 12 – 20 mEq/L</p> <p>Type IV RTA Distal hyperkalemic Decreased aldosterone activity and reduced ammoniogenesis</p>	<p>IV sodium bicarbonate if pH < 7.20</p>

	Typically seen in advanced renal failure, and involves RTAs and retention of organic acids It can also be seen in diarrhea with contraction alkalosis		Associated with hyporeninemic hypoaldosteronism Urine may be acidified Serum HCO ₃ - > 16 mEq/L	
--	--	--	--	--